

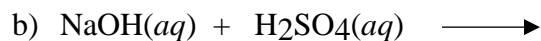
1. Classify each of the following substances as: 1) acid, base, or neutral, and 2) strong or weak.

- a)  $\text{HNO}_3$       b)  $\text{HClO}$       c)  $\text{NH}_3$       d)  $\text{NaNO}_3$       e)  $\text{Ba(OH)}_2$

2. Complete the following equations. Classify the reaction type, write the products and all the equations.



(a)	Reaction type:	Formulas of Products Formed:
(b)	Molecular Equation	
(c)	Total Ionic Equation	
(d)	Net Ionic Equation	



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(b)	Molecular Equation	
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