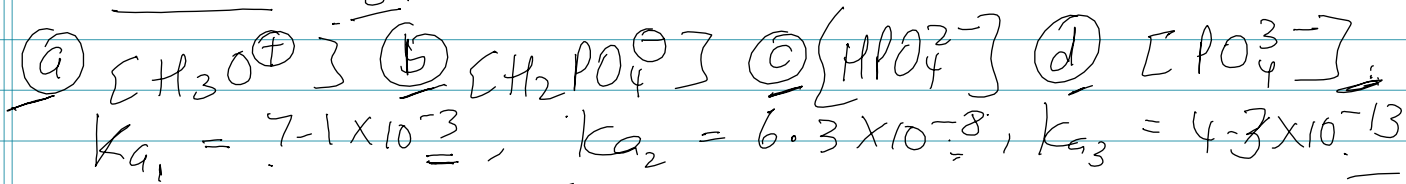


Acid-Base Eq (5) pH of Polyprotic Acid (1)

Calculate the conc. of the following ions in a 5.0M H_3PO_4 soln.



$K_{a1} > K_{a2}$ by 10^5 so most ionization $[H_3O^+]$ is from 1st ionization.

