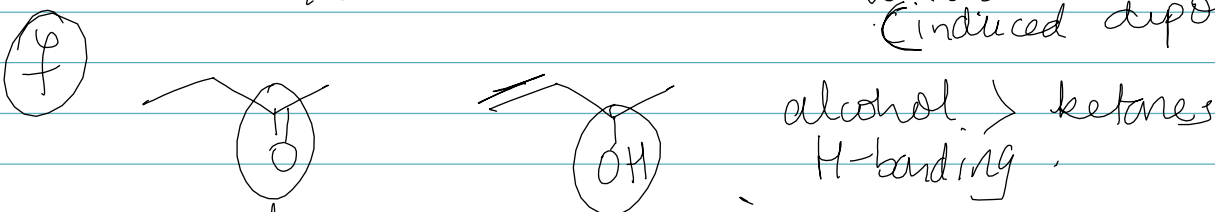
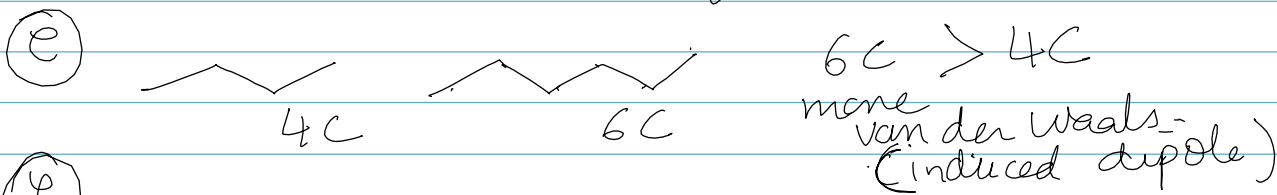
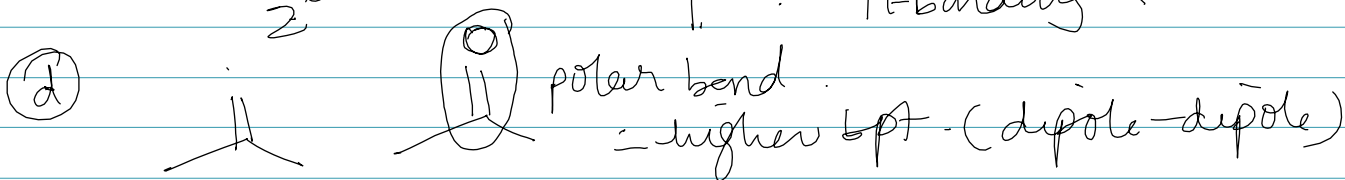
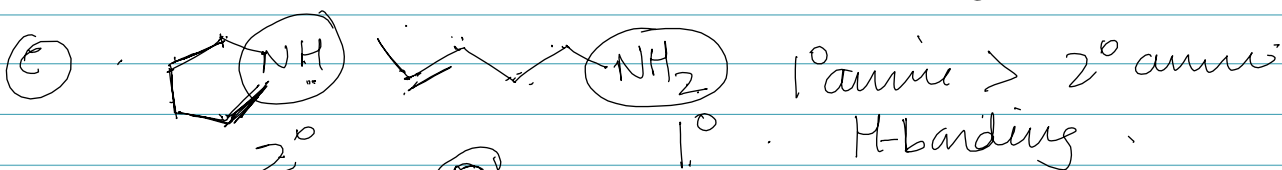
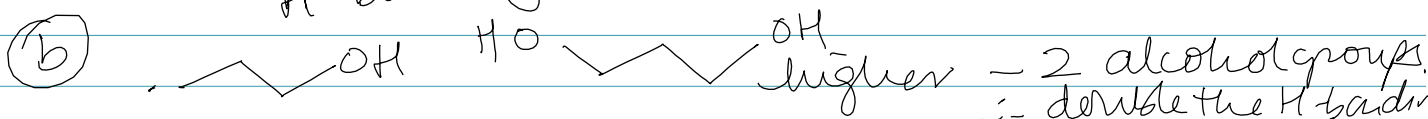
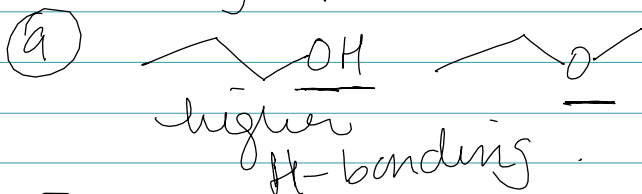


# Intermolecular Forces

Van der Waals (weakest) - no polar bonds.  
 Dipole-Dipole (has polar bonds)  
 H-bonding (strongest) (H & N or F)  
 Ionic compounds - electrostatic force - are polar -

same # of C or same functional group

## 1) Boiling points



## 2) Solubility - like dissolves like.

