- 1) Which of the following reagents might serve as the basis for a simple chemical test that would distinguish between pure 1-hexene and pure hexane?
  - A) Bromine in carbon tetrachloride
- B) Dilute aqueous potassium permanganate

C) Concentrated sulfuric acid

D) All of the above

- E) Answers A) and B) only
- 2) Complete the following reactions with the appropriate starting material, reagents or product. Draw the structures of the chemicals.

b) 
$$H_2C$$
  $CH_3$   $O_3$   $?$   $Zn, AcOH$ 

c) 
$$\frac{\text{H}_2/\text{Pd}}{}$$
 2-methylhexane

d) 
$$\frac{\text{Hot KMnO}_4}{}$$
 2 mols of CH<sub>3</sub>CH<sub>2</sub>COOH

3) Complete the following chart of reactions: