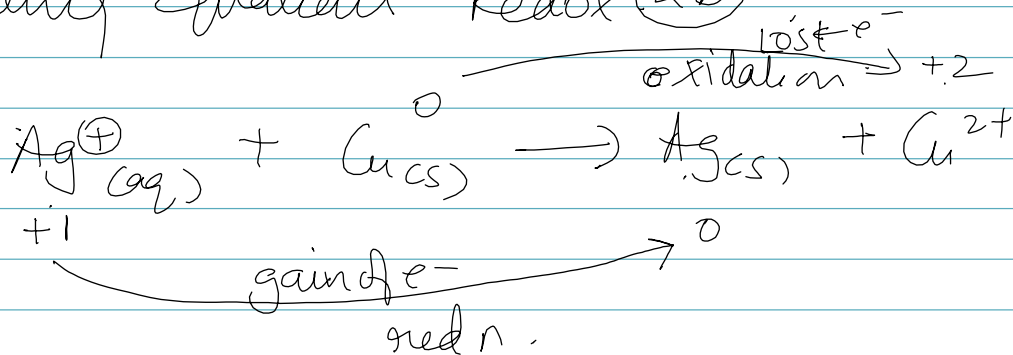
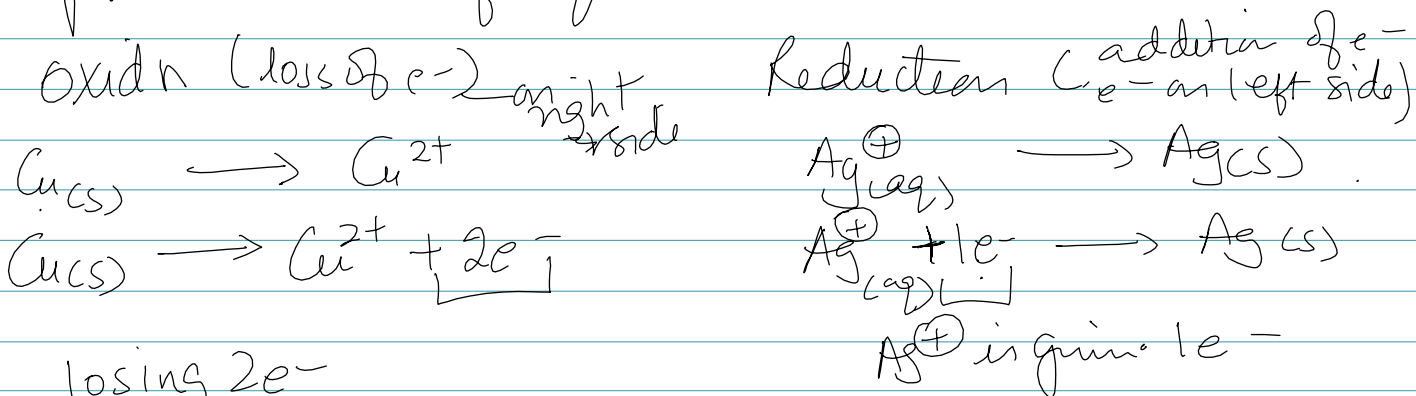


# Writing Equations - Redox (2b)

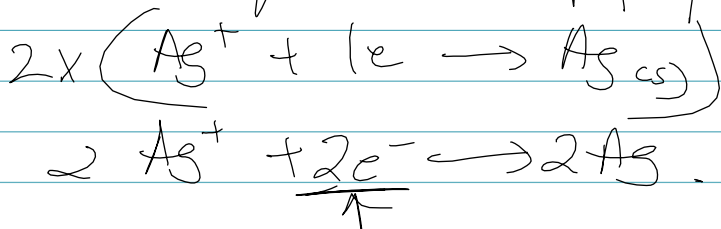
Redox.



separate into half equations.



imbalance of  $e^-$   
 $\text{Cu}$  losing  $2e^-$        $\text{Ag}^{\oplus}$  should gain  $2e^-$   
 take eq. and multiply by 2



Now add equations

