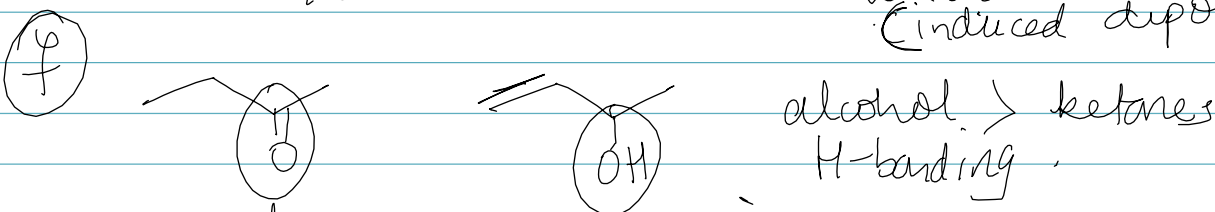
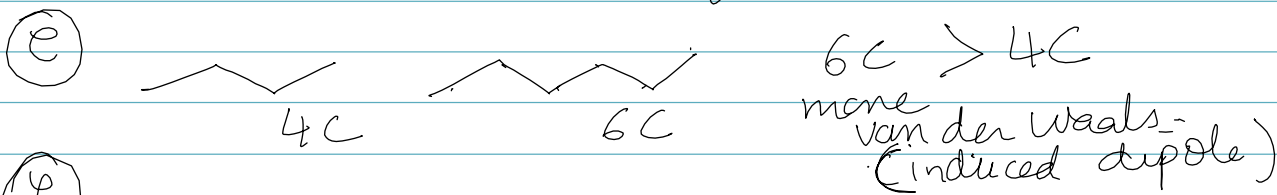
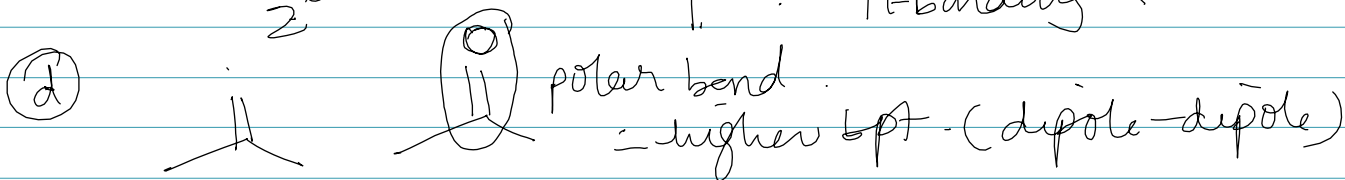
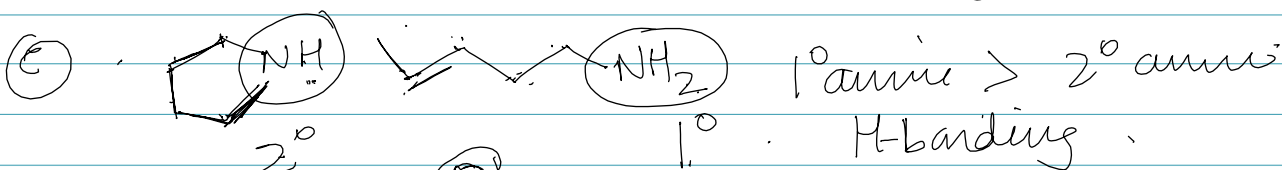
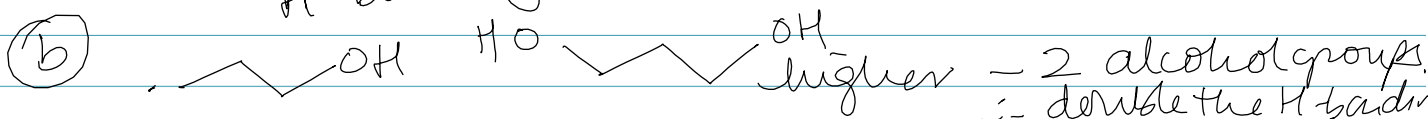
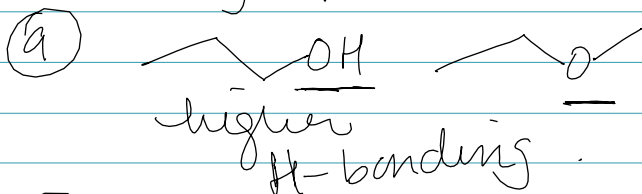


Intermolecular Forces

Van der Waals (weakest) - no polar bonds.
 Dipole-Dipole (has polar bonds)
 H-bonding (strongest) (H & N or F)
 Ionic compounds - electrostatic force - are polar -

same # of C or same functional group

1) Boiling points



2) Solubility - like dissolves like.

