


A Simple Flow Chart Diagram

- 1) Identify the Substrate (it's the one with the leaving group!!)
- 2) identify the Nucleophile/Base (Its the one with the lone pair of electrons and/or the negative charge)
- 3) Define good Nucleophile as any nucleophile with a full negative charge
- 4) Define sterically hindered as secondary or tertiary
- 5) Define strong base as anion on Oxygen, Nitrogen, or Carbon
- 6) Branched 1° = 

Caveat: This will help predict the major product of a particular reactions. This does NOT mean it will be the only product formed, but only will help you consolidate the many and varied conditions you are working with.

